

A Colloidal Model to describe the Effects of Mixing Time on Filler Dispersion in Industrial Nanocomposites

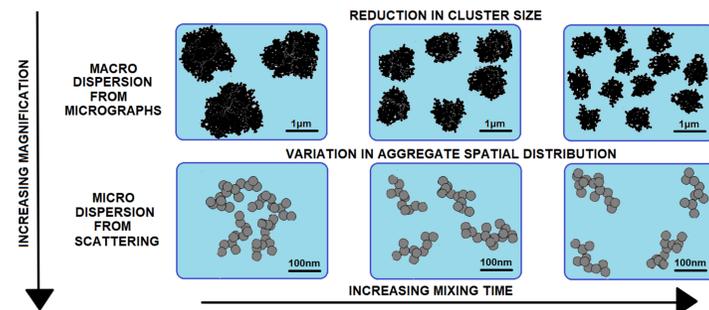
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Abstract

- ❖ Properties of industrially relevant nanocomposites depend on the degree of filler dispersion under high-shear mixing.
- ❖ Conventionally, dispersion is quantified through an index based on the reduction in micron-scale agglomerate size observed in micrographs and bulk electrical conductivity measurements.
- ❖ An alternate nano-scale dispersion technique based on x-ray scattering has been proposed.¹
- ❖ The impact of mixing time on dispersion is investigated taking advantage of the van der Waals equation to describe excluded volume and interaction energy in the dispersion.²
- ❖ An analogy is made between the thermally driven true colloidal dispersions and kinetically accumulated strain in nanocomposites.

Overview



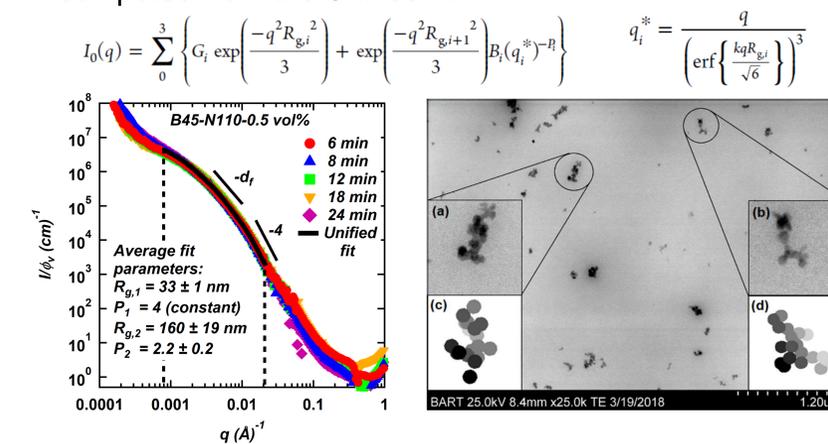
Methods

- Commercial PBD (*Mooney Viscosity* – 38, 45, 54 M.U.) and SBR (*Mooney Viscosity* – 50, 62, 80 M.U.) milled with 6PPD (antioxidant) and varying amount of carbon black reinforcing filler (Vulcan 8 and Vulcan 3) for 6, 8, 12, 18 and 24 mins at 130 °C and 60 rpm.
- Scattering from ~1.2 mm (thk.) flat samples measured at Advanced Photon Source, Argonne National Laboratory using the ultra-small-angle X-ray scattering (USAXS) facility located at the 9 ID beam line, station C.
- Micrographs obtained through TEM in STEM mode from ~80 nm thin sections cryo-cooled below T_g of the nanocomposites

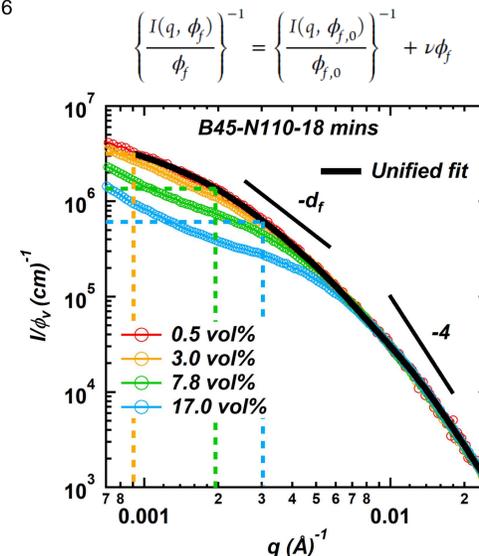
Results

Ultra-small angle X-ray scattering

- ❖ Structural parameters under dilute filler conditions are computed from the Unified fit.^{3,4}



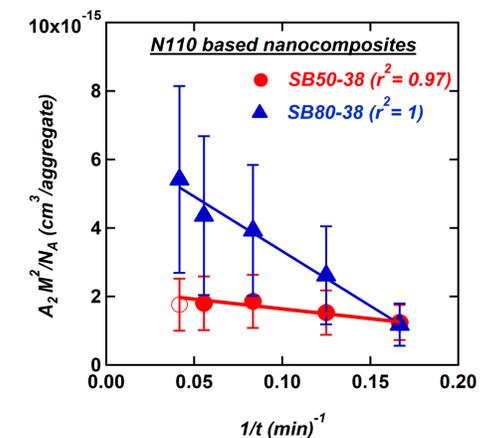
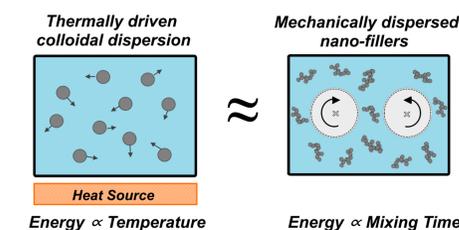
- ❖ Under semi-dilute filler conditions, structural features are screened, and the extent of screening is approximated by RPA.^{5,6}



Kinetic van der Waals model

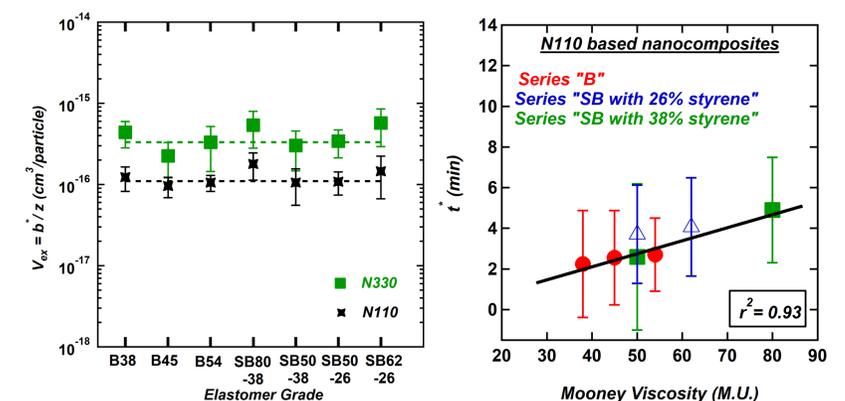
$$A_2 = \frac{v \langle \Delta \rho^2 \rangle^6}{2N_A \rho_f^2}$$

$$A_2(t) = \frac{N_A}{M^2} \left(b^* - \frac{a^*}{t} \right)^2$$



Conclusion

- ❖ The excluded volume depends only on the filler type and seems insensitive to bound rubber.
- ❖ The interaction energy is strongly dependent on viscosity and polymer chemistry.
- ❖ The wetting time for nano-scale incorporation of elastomer into filler can be predicted.



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Acknowledgements

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